

## Chemistry 121 – Chemical Principles II Spring 2009

### I. General Information

<u>Instructor:</u>	<b>Dr. Brian Lamp</b> Office: MG3106	Phone: 785-7251	E-Mail: <a href="mailto:blamp@truman.edu">blamp@truman.edu</a> Internet: <a href="http://www2.truman.edu/~blamp">www2.truman.edu/~blamp</a>
<u>Schedule:</u>	Lecture	MTWF	9:30-10:20 AM MG1090
	Laboratory Section 08	T	11:30 AM -2:20 PM MG1023
	Laboratory Section 01	R	11:30 AM -2:20 PM MG1023
<u>Office Hours:</u>	Office hours are posted outside Dr. Lamp's office and on Dr. Lamp's Website. If you are unable to meet during these hours when in need of help, please arrange a time with the instructor IN ADVANCE.		

### II. Course Philosophy and Goals

Philosophy: Chemistry 120 and 121 are courses that provide a foundation for many other courses in chemistry and biology. Throughout these two courses you will be introduced to a great deal of information. We will not spend a great deal of time on any one topic. Even though we do our best to maintain continuity, at times it may seem like we are skipping from one subject to the next without covering anything in detail. It will help if you remember that you are learning many concepts and techniques that will be very useful in upper-level courses.

Course Goals: In the framework of gaining a better understanding of how science works and the thought processes needed to tackle complex problems, we intend:

1. To learn some fundamental concepts and facts in the field of chemistry.
2. To prepare for more advanced chemistry courses.
3. To learn fundamental skills and safe lab practices.
4. To learn to approach and solve problems in a logical and efficient manner.
5. To enjoy learning about chemistry!

Academic Integrity: Students are expected to abide by the Truman State University Student Conduct Code and complete their coursework, including exams and laboratories using their original words and ideas and properly cite the words and ideas of others. Students caught committing an act of academic misconduct will be subject to the full range of penalties, including failing the course. In every case, the Dean of Student Affairs office and the Vice President for Academic Affairs will be notified.

### III. Required Materials

Textbook: "General Chemistry", 9<sup>th</sup> Ed., Petrucci, Harwood, Herring, and Madura, 2007.

Laboratory Manual: The laboratory manual for Chemical Principles, as well as other lab information is online at [chemlab.truman.edu](http://chemlab.truman.edu).

Laboratory Notebook: A permanently bound laboratory notebook capable of creating duplicate pages is *required*. You must have this notebook by the first lab.

Safety Goggles: All students are required to have departmental approved *safety goggles* for use in the laboratory.

Laboratory Attire: Proper lab attire will be required at **ALL** lab sessions. In addition to safety goggles, you should always attend lab wearing long pants, close-toed shoes, and a shirt that covers the upper body (at least equivalent to a tee-shirt). Inappropriately attired people will not be allowed in lab.

Calculator: You must have a *hand-held scientific calculator* capable of scientific notation for use on homework problems, quizzes, and exams. It will be assumed that you have a calculator for all quizzes and exams. The instructor will not loan calculators to students who have none.

#### IV. Lecture

Coverage: The lecture portion of the course will explore and elaborate on topics presented in the text. It does not make sense for me to repeat all of the material in the text. I will introduce topics and elaborate on points that are particularly essential or challenging. I expect you to read *all* of the chapters that we cover and to ask questions about what you read.

Exams: Four in-class exams worth a total of 400 points will be given. Tentative dates for these exams are:

*Friday, February 6, Wednesday, March 4,  
Friday, April 3, Wednesday, April 29*

A comprehensive final exam will be given Tuesday, May 5 from 9:30 -11:20 AM. The final exam will be worth 200 points.

Homework & Quizzes: It is expected that you will work several practice problems related to each concept we cover in the course. While working in groups to solve problems is very helpful, the result is usually a "group" assignment that does not adequately reflect whether you learned the material. Instead of graded homework, we will have weekly quizzes of 15-20 minutes. These quizzes will be given the last 20 minutes of class each Friday (except for weeks we have an exam). The questions for the quizzes will be taken **directly**, or with slight alteration, from the problems in the text. Your reward for doing your homework conscientiously should be good performance on these quizzes.

Attendance etc. : It is expected that everyone will attend class and participate. Although attendance is not mandatory, roll will regularly be taken. Attendance records are a factor in determining borderline grades at the end of the semester.

Cell Phones/Pagers: Unless you are an emergency responder, all cell phones and pagers must be turned off and stored while in lecture. Failure to do so will result in dismissal from the class session (this includes exams). This will minimize disruption during class and help to maintain Dr. Lamp's good humor!

Make-ups: No make-up exams, quizzes, or labs will be given without prior approval. If you cannot attend a scheduled exam for a **valid, instructor-approved reason**, notify Dr. Lamp IN ADVANCE and an arrangement will be made. No credit will be given for missed labs, exams or quizzes without prior instructor approval. Valid reasons include traveling with a University-sponsored organization, illness, or death in the family. If a sudden, unexpected event occurs that causes your absence, contact the Student Affairs Office and they will contact me. Also, please leave a voice or e-mail message for me and contact me immediately upon your return.

#### IV. Laboratory

Laboratory Philosophy: Experimentation is at the heart of chemistry. The vast majority of the new knowledge in chemistry comes from careful examination and interpretation of experimental results. In order for you to have confidence in your results, it is important that the experiments be executed carefully and properly. The General Chemistry laboratory is designed to teach you lab skills, reinforce concepts from lecture, and show you how to think about scientific problem solving. The importance of lab is reflected in the fact that it counts for about 25% of your final grade.

Lab Notebooks: The purpose of a lab notebook is to document your experiment and enable you or any other worker to understand and duplicate your work. Our goal in using lab notebooks is to teach you the habits that will make you a better scientist in the future. You are required to maintain a permanently bound and numbered lab notebook, capable of making duplicate pages that will be handed in for grading. Detailed formatting guidelines are online at [chemlab.truman.edu](http://chemlab.truman.edu). Before coming to lab, you must familiarize yourself with the theory, techniques, and safety precautions for the experiment. I expect all students to come prepared for lab. If I see students who appear to be reading the lab for the first time during class, I will ban lab manuals from the laboratory.

You must have all the information necessary to do the lab written in your notebook before class. This includes a statement of purpose of the experiment, outline of the procedure that will be followed, and blank data tables if necessary. Failure to come prepared will result in a loss of points. All results and observations must be written **directly** in the notebook, students who make temporary entries on random pieces of paper or paper towels will find them vanishing (the papers, not the students!).

Other guidelines/rules:

- Notebooks must be written in **pen**. Mistakes in the lab notebook should be annotated and crossed out with a single line. Whiteout, etc is not acceptable.
- If labs are not completed in chronological order, be sure that the continuing page is referenced so that the remainder can be found easily.
- All pages must include your name (printed), signature and the date completed.
- Your notebook must have an up-to-date table of contents on the first page.
- Computer-generated tables and graphs are encouraged where appropriate. Such material must be permanently inserted (glue or tape, no staples) into the notebook with a copy on **both** the original and carbon copy notebook page and must not extend off the notebook page.
- If you are working in a group and only one member of the group records data at the time of the experiment, you must indicate in your notebook that the original results were copied from the other person's notebook. Give the page number of the original results.

Lab Grading: Lab performance has a major bearing on your overall course grade. You will be awarded 5 points for arriving at lab with an appropriately prepared lab notebook and 5 points for successfully completing each experiment and turning in a lab report. Several of these reports will be graded for 25-35 points each. The Environmental Chemistry lab will be graded as a 50 point "project".

## VI. Grading

Grade point breakdown: Letter grades will be determined based on the percentage derived from the breakdown shown below.

<u>Grade Point Distribution</u>	
<u>Source</u>	<u>Total Points</u>
4 Exams	400 pts. maximum
9 Quizzes	200 pts. maximum (drop lowest)
Final Exam	200 pts. maximum
Misc. and Group Assignments	100 pts. maximum
10 Experiments	100 pts.
Environmental Chemistry Lab	50 pts.
<u>Lab Reports</u>	<u>125 pts maximum</u>
<b>Total points</b>	<b>1175 pts. maximum</b>

Grading Scale: Awarding of final class grades will be based on the scale below. Percentages will be computed on the basis of total possible points for the semester.

<u>Grade</u>	<u>Percentage</u>
A	90.0-100
B	80.0-89.9
C	70.0-79.9
D	60.0-69.9
F	<60.0

*I reserve the right to lower the grading scale, but it will never be raised.*

Late Hand-ins: **No late work will be accepted.** All assignments will be given a specific deadline, anything turned in after the deadline will earn a score of zero.

## VII. Other Information

- All assignments in lab and lecture *must be written legibly and in a well-organized fashion*. If an answer or work cannot easily be interpreted, no credit will be given.
- All mathematical work and assumptions used when solving a problem, whether on homework, quizzes or exams, *must be shown in order to receive credit for the problem*. Please mark your answers clearly.
- Quiz and exam keys will be posted on the bulletin board outside Dr. Lamp's office. Exam keys, assignments, announcements and other information can be found on Dr. Lamp's CHEM 121 web page at [www2.truman.edu/~blamp](http://www2.truman.edu/~blamp).
- Do not procrastinate!!! Your understanding of lecture material, and grades on homework and exams will be adversely affected by this approach. *It is strongly recommended that you work ahead on reading and homework and participate fully in classroom discussions and problem solving sessions*. The instructor reserves the right to not answer "panic" questions on the day of an exam and will certainly be stingy with time other than his office hours on the day prior to an exam.
- Any student who has a disability that may prevent him/her from fully participating in the class lecture and lab activities must contact the instructor in person as soon as possible to work out solutions to the difficulty. If you have a documented disability, contact the Student Affairs Office immediately to ensure that your needs are properly met. If no contact is made, no concessions will be considered.

***"I'm a great believer in luck, and I find the harder I work the more I have of it."***  
***-Thomas Jefferson***

**VIII. Anticipated Class Calendar: This is only a guide, our schedule may vary slightly.**

Topic	Chapter
A Atomic Structure & Quantum Theory	8
B Atomic Electron Config. & Periodicity	8/9
C Bonding and Molecular Structure	10
D Orbitals and Chemical Bonding	11
E Organic and Carbon Chemistry	26
F Structure, Reactivity & Acid-Base	16/17
G IM Forces, Liquids and Solids	12
H Solutions and Their Behavior	13

Topic	Chapter
I Thermodynamics Review	7/19
J Chemical Equilibrium Review	16-18
K Electron-Transfer Reactions	20
L Chemical Kinetics Review	14
M <i>Main Group Elements</i>	21/22
N Transition Metals & Coord. Chem.	23/24
O Nuclear Chemistry	25

**January**

	Monday	Tuesday	Wednesday	Thursday	Friday	Lab
Date	12	13	14	15	16	No Lab
Topic	A	A	A		A	No Lab
Date	19	20	21	22	23	Qual. Anal. Cations
Topic	MLK Holiday	A/B	B		B/Quiz	
Date	26	27	28	29	30	Qual. Anal. Anions
Topic	C	C	C		C/Quiz	

**February**

	Monday	Tuesday	Wednesday	Thursday	Friday	Lab
Date	2	3	4	5	6	Qual. Anal. Unknowns
Topic	C	C	D		Exam	
Date	9	10	11	12	13	Molec. Models 1
Topic	D	D	D		D/Quiz	
Date	16	17	18	19	20	No Lab
Topic	D	D/E	E	Univ. Conf.	E/Quiz	
Date	23	24	25	26	27	Electroneg. by NMR
Topic	F	F	F		F/Quiz	

**March**

	Monday	Tuesday	Wednesday	Thursday	Friday	Lab
Date	2	3	4	5	6	Synthesis Coord. Cpds
Topic	G	G	Exam		G	
Date	9	10	11	12	13	No Lab
Topic			Mid Term Brk.			
Date	16	17	18	19	20	Character. Coord. Cpds
Topic	H	H	H		I/Quiz	
Date	23	24	25	26	27	Molec. Models 2
Topic	I/J	J	J/K		K/Quiz	
Date	30	31				Env. Chem. Planning*
Topic	K	K				

**April**

	Monday	Tuesday	Wednesday	Thursday	Friday	Lab
Date			1	2	3	Env. Chem. Planning
Topic			L		Exam	
Date	6	7	8	9	10	No Lab
Topic	L	SRC	M		M/Quiz	
Date	13	14	15	16	17	Env. Chem. Planning
Topic	Spring Break	M	N		N/Quiz	
Date	20	21	22	23	24	Env. Chem.
Topic	N	N/O	O		O/Quiz	
Date	27	28	29	30	May 1	Presentations/ Check Out
Topic	O	O	Exam		Review	

**May**

	Monday	Tuesday	Wednesday	Thursday	Friday	Lab
Date	4	5	6	7	8	No Lab
Topic		Final Exam	Finals Week			